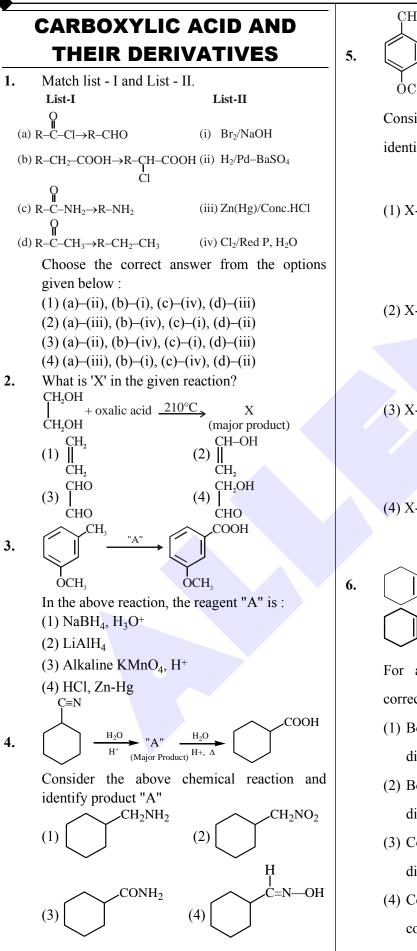
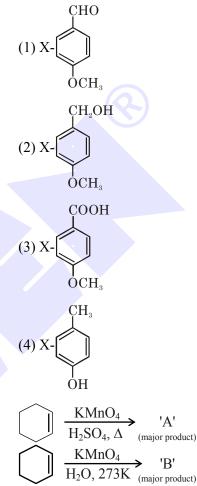
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$$\overset{\operatorname{CH}_3}{\underset{\operatorname{OCH}_3}{\longleftarrow}} \overset{\operatorname{Alkaline} \operatorname{KMnO}_4}{\underset{\operatorname{H}^+}{\longrightarrow}} "X"$$

Considering the above chemical reaction, identify the product "X" :

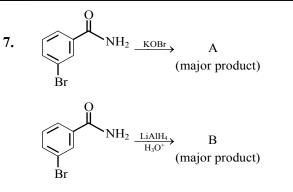


For above chemical reactions, identify the correct statement from the following:

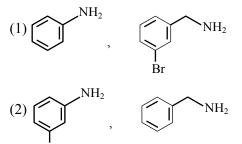
- Both compound 'A' and compound 'B' are dicarboxylic acids
- (2) Both compound 'A' and compound 'B' are diols
- (3) Compound 'A' is diol and compound 'B' is dicarboxylic acid
- (4) Compound 'A' is dicarboxylic acid and compound 'B' is diol

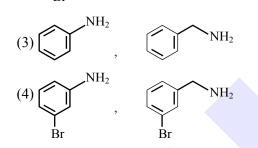
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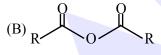


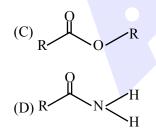
In the above reactions, product A and product B respectively are :





$$\mathbf{8.} \quad \mathbf{(A)} \overset{\mathbf{O}}{\underset{\mathbf{R}}{\overset{\mathbf{O}}}{\overset{\mathbf{O}}}{\overset{\mathbf{O}}{\overset{\mathbf{O}}{\overset{\mathbf{O}}{\overset{\mathbf{O}}{\overset{\mathcal{O}}{\overset{\mathcal{O}}}{\overset{\mathcal{O}}{\overset{\mathcal{O}}{\overset{\mathcal{O}}}{\overset{\mathcal{O}}{\overset{\mathcal{O}}{\overset{\mathcal{O}}{\overset{\mathcal{O}}{\overset{\mathcal{O}}}{\overset{\mathcal{O}}}}{\overset{\mathcal{O}$$





The **correct** order of their reactivity towards hydrolysis at room temperature is :

- (1) (A) > (B) > (C) > (D)(2) (D) > (A) > (B) > (C)(3) (D) > (B) > (A) > (C)(4) > (C)
- (4) (A) > (C) > (B) > (D)

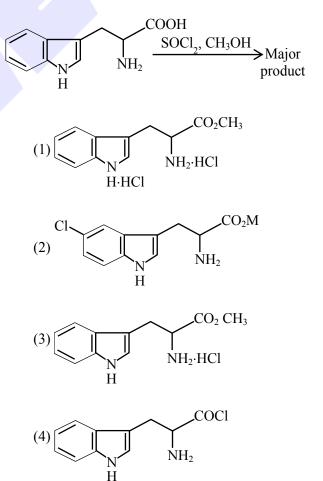


9.

Maleic anhydride

Maleic anhydride can be prepared by :

- (1) Heating trans-but-2-enedioic acid
- (2) Heating cis-but-2-enedioic acid
- (3) Treating cis-but-2-enedioic acid with alcohol and acid
- (4) Treating trans-but-2-enedioic acid with alcohol and acid
- **10.** The major product formed in the following reaction is :



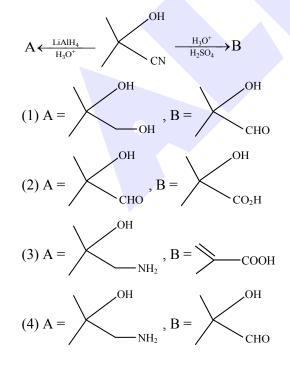
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Carboxylic Acid and Their Derivatives 3

11. The major product of the following reaction is :

ÇH₃ (i) alcoholic NH₃ -Cl (ii) NaOH, Br₂ CH₃-CH-CH₂-CH₂-C-➤ Major (iii) NaNO₂,HCl product (iv) H₂O Br (1) CH₃-CH-CH-CH₂OH CH_3 (2) CH_3 –CH– CH_2 – CH_2 – CH_2OH ĊH₃ (3) CH₃–CH–CH₂–CH₂OH ĊH₃ (4) CH₃–CH–CH₂–CH₂–Cl ĊH₃

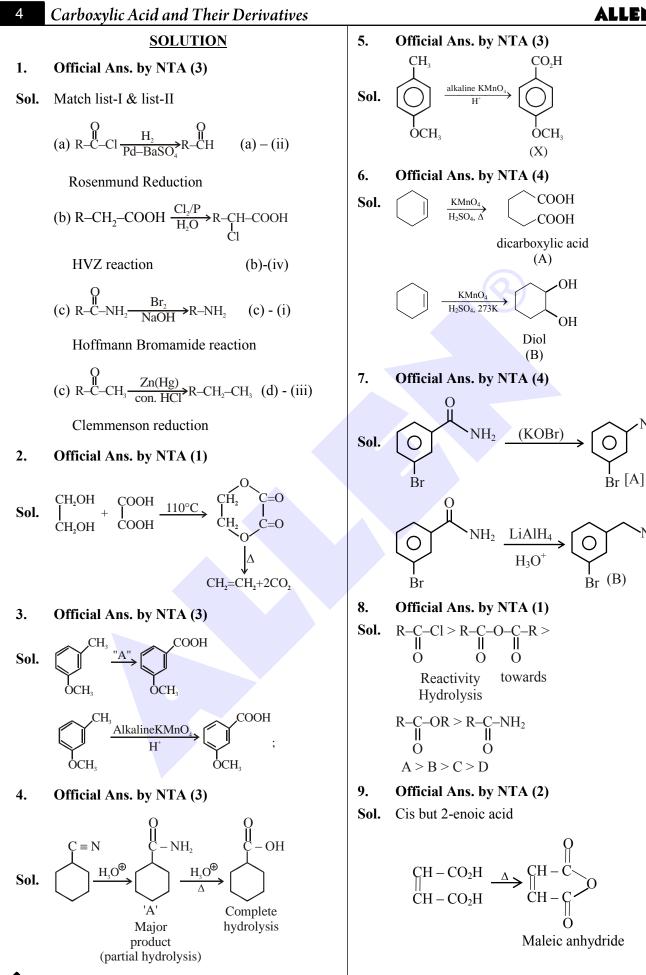
- **12.** Which one of the following reactions will **not** yield propionic acid?
 - (1) $CH_3CH_2COCH_3 + OI^-/H_3O^+$
 - (2) $CH_3CH_2CH_3 + KMnO_4$ (Heat), OH^-/H_3O^+
 - $(3) CH_3CH_2CCl_3 + OH^-/H_3O^+$
 - (4)CH₃CH₂CH₂Br + Mg, CO₂ dry ether/H₃O⁺
- **13.** The major products A and B in the following set of reactions are :



14. In the following sequence of reactions a compound A, (molecular formula $C_6H_{12}O_2$) with a straight chain structure gives a C_4 carboxylic acid. A is :

$$A \xrightarrow{\text{LiAlH}_4} B \xrightarrow{\text{Oxidation}} C_4 - \text{carboxylic acid}$$
(1) CH₃-CH₂-COO-CH₂-CH₂-CH₃
OH
(2) CH₃-CH₂-CH-CH₂-O-CH=CH₂
(3) CH₃-CH₂-CH₂-COO-CH₂-CH₃
(4) CH₃-CH₂-CH₂-O-CH = CH-CH₂-OH

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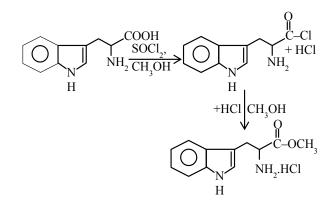
 $\sim NH_2$

·NH₂

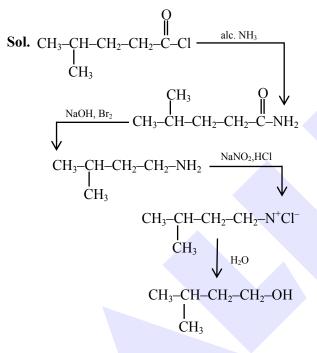
13.

Official Ans. by NTA (3) 10.

Sol.



11. Official Ans. by NTA (3)



12. Official Ans. by NTA (4)

All gives propanoic acid as product but option 4 Sol.

gives butanoic as product

$$CH_{3}CH_{2}CH_{2}Br \xrightarrow{Mg} CH_{3}CH_{2}CH_{2}MgBr$$

$$CH_{3}-CH_{2}-CH_{2}-C-OMgBr$$

$$H_{3}O^{+} O$$

$$CH_{3}-CH_{2}-CH_{2}-C-OH$$
Butanoic acid

- Official Ans. by NTA (3) ЮH ЮН Sol. LiAlH₄ $\mathrm{H_3O}^+$ H₃O C-OH $\dot{C}H_2 - NH_2$ ℃-|| 0 H_2SO_4 -CO₂H 14. Official Ans. by NTA (3) **Sol.** CH_3 - CH_2 - CH_2 - CH_2 - CH_3 (A) $[C_6H_{12}O_2]$ Ö (1) LiAlH₄ $(2) H_3O^+$
 - $CH_3-CH_2-CH_2-CH_2-OH + CH_3 CH_2-OH$ **(B)** [O]
 - [C₄ carboxylic acid] CH₃-CH₂-CH₂-C-OH 0