## AROMATIC COMPOUND

1. ' A ' and ' B ' in the following reactions are :

(1) (A)

(B)

(2) (A) :

(B)

(3) (A)

(B)

(4) (A)

(B)

2. In the following reaction the reason why meta-nitro product also formed is :

(1) low temperature
(2) $-\mathrm{NH}_{2}$ group is highly meta-directive
(3) Formation of anilinium ion
(4) $-\mathrm{NO}_{2}$ substitution always takes place at meta-position
3. Which of the following compound gives pink colour on reaction with phthalic anhydride in conc. $\mathrm{H}_{2} \mathrm{SO}_{4}$ followed by treatment with NaOH ?
(1)

(2)

(3)

(4)

4. What is the correct sequence of reagents used for converting nitrobenzene into $m$-dibromobenzene ?

(1)

$(2) \xrightarrow{\mathrm{Br}_{2} / \mathrm{Fe}} / \xrightarrow{\mathrm{Sn} / \mathrm{HCl}} / \xrightarrow{\mathrm{NaNO}_{2} / \mathrm{HCl}} / \xrightarrow{\mathrm{CuBr} / \mathrm{HBr}}$
(3)

(4) $\xrightarrow{\mathrm{Sn} / \mathrm{HCl}} / \xrightarrow{\mathrm{Br}_{2}} / \xrightarrow{\mathrm{NaNO}_{2}} / \xrightarrow{\mathrm{NaBr}}$
5. The correct order of the following compounds showing increasing tendency towards nucleophilic substitution reaction is :-

(i)

(iii)

(iv)
(1) (iv) $<$ (iii) $<$ (ii) $<$ (i)
(2) (iv) $<$ (i) $<$ (ii) $<$ (iii)
(3) (iv) < (i) < (iii) < (ii)
(4) (i) < (ii) < (iii) < (iv)
6. The diazonium salt of which of the following compounds will form a coloured dye on reaction with $\beta$-Naphthol in NaOH ?
(1)

(2)

(3)

(4)

7. Which of the following reaction/s will not give p-aminoazobenzene?
(A)

(B)

(C)

(1) A only
(2) B only
(3) C only
(4) A and B
8. Identify A in the given chemical reaction.

(1)

(2)

(3)

(4)

9. Consider the following chemical reaction.
$\mathrm{CH} \equiv \mathrm{CH} \xrightarrow[(2) \mathrm{CO}, \mathrm{HCl}, \mathrm{AlCl}_{3}]{\text { (1) Red hot Fe tube, } 873 \mathrm{~K}}$ Product
The number of $\mathrm{sp}^{2}$ hybridized carbon atom(s) present in the product is $\qquad$ .
10. The major product of the following reaction is:

(1)

(2)

(3)

(4)

11. The correct sequence of reagents used in the preparation of 4-bromo-2-nitroethyl benzene from benzene is :
(1) $\mathrm{HNO}_{3} / \mathrm{H}_{2} \mathrm{SO}_{4}, \mathrm{Br}_{2} / \mathrm{AlCl}_{3}, \mathrm{CH}_{3} \mathrm{COCl} / \mathrm{AlCl}_{3}$, $\mathrm{Zn}-\mathrm{Hg} / \mathrm{HCl}$
(2) $\mathrm{Br}_{2} / \mathrm{AlBr}_{3}, \mathrm{CH}_{3} \mathrm{COCl} / \mathrm{AlCl}_{3}, \mathrm{HNO}_{3} / \mathrm{H}_{2} \mathrm{SO}_{4}$, $\mathrm{Zn} / \mathrm{HCl}$
(3) $\mathrm{CH}_{3} \mathrm{COCl} / \mathrm{AlCl}_{3}, \mathrm{Br}_{2} / \mathrm{AlBr}_{3}, \mathrm{HNO}_{3} / \mathrm{H}_{2} \mathrm{SO}_{4}$, $\mathrm{Zn} / \mathrm{HCl}$
(4) $\mathrm{CH}_{3} \mathrm{COCl} / \mathrm{AlCl}_{3}, \mathrm{Zn}-\mathrm{Hg} / \mathrm{HCl}, \mathrm{Br}_{2} / \mathrm{AlBr}_{3}$, $\mathrm{HNO}_{3} / \mathrm{H}_{2} \mathrm{SO}_{4}$
12. 



(A) (B)

Correct statement about the given chemical reaction is :
(1) $-\stackrel{\mathrm{N}}{\mathrm{H}_{2}}$ group is ortho and para directive, so product ( B ) is not possible.
(2) Reaction is possible and compound (B) will be the major product.
(3) The reaction will form sulphonated product instead of nitration.
(4) Reaction is possible and compound (A) will be major product.
13. Identify the major products $A$ and $B$ respectively in the following reactions of phenol.
(B)

(1)
 and

(2)

(3)
 and

(4)

14. Match List-I with List-II

## List-I

(a)


(b)


(c) $2 \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{Cl}+2 \mathrm{Na} \xrightarrow{\text { E.her }} \mathrm{C}_{2} \mathrm{H}_{5}-\mathrm{C}_{2} \mathrm{H}_{5}+2 \mathrm{NaCl}$
(d) $2 \mathrm{C}_{6} \mathrm{H}_{5} \mathrm{Cl}+2 \mathrm{Na} \xrightarrow{\text { Ether }} \mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{C}_{6} \mathrm{H}_{5}+2 \mathrm{NaCl}$

## List-II

(i) Wurtz reaction
(ii) Sandmeyer reaction
(iii)Fittig reaction
(iv) Gatterman reaction

Choose the correct answer from the options given below :
(1) (a) $\rightarrow$ (iii), (b) $\rightarrow$ (i), (c) $\rightarrow$ (iv), (d) $\rightarrow$ (ii)
(2) (a) $\rightarrow$ (ii), (b) $\rightarrow$ (i), (c) $\rightarrow$ (iv), (d) $\rightarrow$ (iii)
(3) (a) $\rightarrow$ (ii), (b) $\rightarrow$ (iv), (c) $\rightarrow$ (i), (d) $\rightarrow$ (iii)
(4) (a) $\rightarrow$ (iii), (b) $\rightarrow$ (iv), (c) $\rightarrow$ (i), (d) $\rightarrow$ (ii)
15.


In the above chemical reaction, intermediate " X " and reagent/condition " A " are :
(1) ${ }^{X}$

(2) $x$
 ; $\mathrm{A}-\mathrm{H}_{2} \mathrm{O} / \Delta$
(3) $X$

; $\mathrm{A}-\mathrm{H}_{2} \mathrm{O} / \Delta$
(4) $X$
 ; $\mathrm{A}-\mathrm{H}_{2} \mathrm{O} / \mathrm{NaOH}$
16.


The above reaction requires which of the following reaction conditions?
(1) $573 \mathrm{~K}, \mathrm{Cu}, 300 \mathrm{~atm}$
(2) $623 \mathrm{~K}, \mathrm{Cu}, 300 \mathrm{~atm}$
(3) $573 \mathrm{~K}, 300 \mathrm{~atm}$
(4) $623 \mathrm{~K}, 300 \mathrm{~atm}$
17.


In the above reaction, 3.9 g of benzene on nitration gives 4.92 g of nitrobenzene. The percentage yield of nitrobenzene in the above reaction is $\qquad$ \%. (Round off to the Nearest Integer).
(Given atomic mass : C : $12.0 \mathrm{u}, \mathrm{H}: 1.0 \mathrm{u}, \mathrm{O}:$
$16.0 \mathrm{u}, \mathrm{N}: 14.0 \mathrm{u})$
18.


In the above reaction, the structural formula of
(A), "X" and "Y" respectively are :
(1)

(2)


(3)


(4)




19.


Considering the above reaction, X and Y respectively are :
(1)

(2)

(3)
 and

(4)

20.


Consider the given reaction, percentage yield of :
(1) $\mathrm{C}>\mathrm{A}>$ B
(2) B $>$ C $>$ A
(3) $\mathrm{A}>\mathrm{C}>\mathrm{B}$
(4) $\mathrm{C}>\mathrm{B}>\mathrm{A}$
21.


Consider the above reaction where 6.1 g of benzoic acid is used to get 7.8 g of m-bromo benzoic acid. The percentage yield of the product is $\qquad$ .
(Round off to the Nearest integer)
[Given: Atomic masses : $\mathrm{C}=12.0 \mathrm{u}, \mathrm{H}: 1.0 \mathrm{u}$, $\mathrm{O}: 16.0 \mathrm{u}, \mathrm{Br}=80.0 \mathrm{u}]$
22. The correct structure of Rhumann's Purple, the compound formed in the reaction of ninhydrin with proteins is :
(1)

(2)

(3)

(4)

23. Benzene on nitration gives nitrobenzene in presence of $\mathrm{HNO}_{3}$ and $\mathrm{H}_{2} \mathrm{SO}_{4}$ mixture, where :
(1) both $\mathrm{H}_{2} \mathrm{SO}_{4}$ and $\mathrm{HNO}_{3}$ act as a bases
(2) $\mathrm{HNO}_{3}$ acts as an acid and $\mathrm{H}_{2} \mathrm{SO}_{4}$ acts as a base
(3) both $\mathrm{H}_{2} \mathrm{SO}_{4}$ and $\mathrm{HNO}_{3}$ act as an acids
(4) $\mathrm{HNO}_{3}$ acts as a base and $\mathrm{H}_{2} \mathrm{SO}_{4}$ acts as an acid
24.



Consider the above reaction, compound B is :
(1)

(2)

(3)

(4)

25.


Major product P of above reaction, is :
(1)

(2)

(3)

(4)

26. Which of the following compounds will provide a tertiary alcohol on reaction with excess of $\mathrm{CH}_{3} \mathrm{MgBr}$ followed by hydrolysis?
(1)

(2)

(3)

(4)

27.


In the chemical reactions given above A and B respectively are :
(1) $\mathrm{H}_{3} \mathrm{PO}_{2}$ and $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{Cl}$
(2) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{OH}$ and $\mathrm{H}_{3} \mathrm{PO}_{2}$
(3) $\mathrm{H}_{3} \mathrm{PO}_{2}$ and $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{OH}$
(4) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{Cl}$ and $\mathrm{H}_{3} \mathrm{PO}_{2}$
28. An organic compound $A\left(\mathrm{C}_{6} \mathrm{H}_{6} \mathrm{O}\right)$ gives dark green colouration with ferric chloride. On treatment with $\mathrm{CHCl}_{3}$ and KOH , followed by acidification gives compound B . Compound B can also be obtained from compound C on reaction with pyridinium chlorochromate (PCC). Identify A, B and C .
(1)

(2)

(3)



(4)
 $\mathrm{B}=$


29. Which one of the following reactions does not occur?
(1)

(2)

(3)

(4)

30.


The given reaction can occur in the presence of :
(a) Bromine water
(b) $\mathrm{Br}_{2}$ in $\mathrm{CS}_{2}, 273 \mathrm{~K}$
(c) $\mathrm{Br}_{2} / \mathrm{FeBr}_{3}$
(d) $\mathrm{Br}_{2}$ in $\mathrm{CHCl}_{3}, 273 \mathrm{~K}$

Choose the correct answer from the options given below :
(1) (b) and (d) only
(2) (a) and (c) only
(3) (b), (c) and (d) only
(4) (a), (b) and (d) only
31. Which one of the products of the following reactions does not react with Hinsberg reagent to form sulphonamide?
(1)

(2)

(3)

(4)

32. The correct decreasing order of densities of the following compounds is :

(A)

(B)

(C)

(D)
(1) (D) $>$ (C) $>$ (B) $>$ (A)
(2) (C) $>$ (D) $>($ A $)>($ B $)$
(3) (C) $>$ (B) $>$ (A) $>$ (D)
(4) (A) $>$ (B) $>$ (C) $>$ (D)
33. $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{NO}_{2} \xrightarrow{\mathrm{Sn}+\mathrm{HCl}^{2}}$ "A" $\xrightarrow[\mathrm{H}^{\oplus}]{\mathrm{C}_{6} \mathrm{H}_{5} \stackrel{\oplus}{\mathrm{~N}_{2}} \stackrel{\ominus}{\mathrm{C}}} \underset{\text { (Yellow }}{\mathrm{P}}$ coloured compound)
Consider the above reaction, the Product " P " is :
(1)

(2)

(3)

(4)

34. The correct sequence of correct reagents for the following transformation is :-

(1) (i) $\mathrm{Fe}, \mathrm{HCl}$
(ii) $\mathrm{Cl}_{2}, \mathrm{HCl}$,
(iii) $\mathrm{NaNO}_{2}, \mathrm{HCl}, 0^{\circ} \mathrm{C}$ (iv) $\mathrm{H}_{2} \mathrm{O} / \mathrm{H}^{+}$
(2) (i) $\mathrm{Fe}, \mathrm{HCl}$
(ii) $\mathrm{NaNO}_{2}, \mathrm{HCl}, 0^{\circ} \mathrm{C}$
(iii) $\mathrm{H}_{2} \mathrm{O} / \mathrm{H}^{+}$
(iv) $\mathrm{Cl}_{2}, \mathrm{FeCl}_{3}$
(3) (i) $\mathrm{Cl}_{2}, \mathrm{FeCl}_{3}$
(ii) $\mathrm{Fe}, \mathrm{HCl}$
(iii) $\mathrm{NaNO}_{2}, \mathrm{HCl}, 0^{\circ} \mathrm{C}$ (iv) $\mathrm{H}_{2} \mathrm{O} / \mathrm{H}^{+}$
(4) (i) $\mathrm{Cl}_{2}, \mathrm{FeCl}_{3}$
(ii) $\mathrm{NaNO}_{2}, \mathrm{HCl}, 0^{\circ} \mathrm{C}$
(iii) $\mathrm{Fe}, \mathrm{HCl}$
(iv) $\mathrm{H}_{2} \mathrm{O} / \mathrm{H}^{+}$
35. The correct options for the products $\mathbf{A}$ and $\mathbf{B}$ of the following reactions are :

(1) $\mathbf{A}=$


B=

(2) $\mathbf{A}=$
 , $\mathbf{B}=$

(3)


(4)

36. The correct sequential addition of reagents in the preparation of 3-nitrobenzoic acid from benzene is:
(1) $\mathrm{Br}_{2} / \mathrm{AlBr}_{3}, \mathrm{HNO}_{3} / \mathrm{H}_{2} \mathrm{SO}_{4}, \mathrm{Mg}$ ether, $\mathrm{CO}_{2}, \mathrm{H}_{3} \mathrm{O}^{+}$
(2) $\mathrm{Br}_{2} / \mathrm{AlBr}_{3}, \mathrm{NaCN}, \mathrm{H}_{3} \mathrm{O}^{+}, \mathrm{HNO}_{3} / \mathrm{H}_{2} \mathrm{SO}_{4}$
(3) $\mathrm{Br}_{2} / \mathrm{AlBr}_{3}, \mathrm{HNO}_{3} / \mathrm{H}_{2} \mathrm{SO}_{4}, \mathrm{NaCN}, \mathrm{H}_{3} \mathrm{O}^{+}$
(4) $\mathrm{HNO}_{3} / \mathrm{H}_{2} \mathrm{SO}_{4}, \mathrm{Br}_{2} / \mathrm{AlBr}_{3}$, Mg/ether, $\mathrm{CO}_{2}, \mathrm{H}_{3} \mathrm{O}^{+}$
37. Which one of the following phenols does not give colour when condensed with phthalic anhydride in presence of conc. $\mathrm{H}_{2} \mathrm{SO}_{4}$ ?
(1)

(2)

(3)

(4)

38.


Consider the given reaction, the product $\mathbf{A}$ is:
(1)

(2)

(3)

(4)

39. The structure of product $C$, formed by the following sequence of reactions is :

(1)

(2)

(3)

(4)

40. The total number of reagents from those given below, that can convert nitrobenzene into aniline is $\qquad$ . (Integer answer)
I. $\mathrm{Sn}-\mathrm{HCl}$
II. $\mathrm{Sn}-\mathrm{NH}_{4} \mathrm{OH}$
III. $\mathrm{Fe}-\mathrm{HCl}$
IV. $\mathrm{Zn}-\mathrm{HCl}$
V. $\mathrm{H}_{2}-\mathrm{Pd}$
VI. $\mathrm{H}_{2}$ - Raney Nickel
41. The structures of $\mathbf{A}$ and $\mathbf{B}$ formed in the following reaction are : $\left[\mathrm{Ph}=-\mathrm{C}_{6} \mathrm{H}_{5}\right]$

(1)

(2)

(3)

(4)



42. For the following :

(1)

(2)

(3)

(4)

43. Identify correct $\mathbf{A}, \mathbf{B}$ and $\mathbf{C}$ in the reaction sequence given below :

(1)

(2)

(3)

(4) A

44. Identify A in the following reaction.

(1)

(2)

(3)

(4)


## SOLUTION

1. Official Ans. by NTA (3)

Sol.

2. Official Ans. by NTA (3)

Sol.

(Anilinium ion)
Aniline on protonation gives anilinium ion which is meta directing. So considerable amount of meta product is formed.
3. Official Ans. by NTA (1)

Sol.

4. Official Ans. by NTA (2)

Sol. Correct sequence of reagents for the following conversion.


5. Official Ans. by NTA (4)

Sol. For nucleophile substitution in aromatic halides

(i)

(ii)

(iii)

(iv)

Correct order is :
(i) < (ii) < (iii) < (iv)

More No. of $\mathrm{NO}_{2}$ substituted aromatic halide, increase the rate of nucleophile substitution reaction in aromatic halides.
6. Official Ans. by NTA (3)

Sol.

$1^{\circ}$ aromatic amine
7. Official Ans. by NTA (2)

Sol. In basic or neutral medium $\mathrm{N}-\mathrm{N}$ coupling favourable while in slightly acidic medium $\mathrm{C}-\mathrm{N}$ coupling favourable.
(A)

(B)

(C)


8. Official Ans. by NTA (4)

Sol.

$\mathrm{Mo}_{2} \mathrm{O}_{3}$ at 773 K temperature and 10-20-atm pressure is aromatising agent.
9. Official Ans. by NTA (7)

Sol.


In benzaldehyde total number of $\mathrm{sp}^{2} \mathrm{C}^{\prime}$ are 7 .
10. Official Ans. by NTA (3)

Sol.


11. Official Ans. by NTA (4)

Sol.


$\mathrm{Br}_{2} / \mathrm{AlBr}_{3}$


12. Official Ans. by NTA (4)

Sol.


Due to formation of anilinium ion by acid base
reaction m -product is form as considerable amount.

13. Official Ans. by NTA (2)

Sol. (B)


14. Official Ans. by NTA (3)

Sol. (a) $\rightarrow$ (ii) Sand Meyer reaction
(b) $\rightarrow$ (iv) Gatterman reaction
(c) $\rightarrow$ (i) Wurtz reaction
(d) $\rightarrow$ (iii) Fittig reaction
(a) $\rightarrow$ (ii),
(b) $\rightarrow$ (iv),
(c) $\rightarrow$ (i),
(d) $\rightarrow$ (iii)
15. Official Ans. by NTA (3)

Sol.

16. Official Ans. by NTA (2)

Sol. Positive zero error $=0.2 \mathrm{~mm}$
Main scale reading $=8.5 \mathrm{~cm}$
Vernier scale reading $=6 \times 0.01=0.06 \mathrm{~cm}$
Final reading $=8.5+0.06-0.02=8.54 \mathrm{~cm}$
17. Official Ans. by NTA (80)

Sol.


1 mole
1 mole
78 gm
123 gm
3.9 gm

$$
\frac{123}{78} \times 3.9=6.15 \mathrm{gm}
$$

But actual amount of nitrobenzene formed is 4.92 gm and hence.

Percentage yield $=\frac{4.92}{6.15} \times 100=80 \%$
18. Official Ans. by NTA (1)

Sol.

19. Official Ans. by NTA (2)

Sol.

(Y)

Major product
20. Official Ans. by NTA (4)

Sol.


Aniline

\% yield order $\Rightarrow \mathrm{C}>\mathrm{B}>\mathrm{A}$
21. Official Ans. by NTA (78)

Sol. Moles of Benzoic acid $=\frac{6.1}{122}$
$=$ moles of m-bromobenzoic acid
So, weight of m-bromobenzoic acid
$=\frac{6.1}{122} \times 201 \mathrm{gm}$
$=10.05 \mathrm{gm}$
$\%$ yield $=\frac{\text { Actual weight }}{\text { Theoretical weight }} \times 100$
$=\frac{7.8}{10.05} \times 100$
$=77.61 \%$
22. Official Ans. by NTA (4)

Sol.



Ninhydrin Test
23. Official Ans. by NTA (4)

Sol. Reagent for nitration of Benzene

$$
\begin{aligned}
& \underset{\text { (Acid) }}{\mathrm{H}_{2} \mathrm{SO}_{4}}+\underset{\text { (Base) }}{\mathrm{HNO}_{3}} \rightleftarrows \stackrel{\mathrm{HSO}_{4}^{\ominus}}{\oplus}+\mathrm{H}_{2} \stackrel{\oplus}{\mathrm{~N}} \mathrm{O}_{3} \\
& \mathrm{H}_{2} \stackrel{\oplus}{\mathrm{NO}_{3}} \rightleftarrows \mathrm{H}_{2} \mathrm{O}+\stackrel{\oplus}{\mathrm{NO}_{2}}
\end{aligned}
$$


24. Official Ans. by NTA (3)


25. Official Ans. by NTA (4)

Sol.



26. Official Ans. by NTA (1)

Sol.




Phenolic - OH group and tertiary alcohol is present thus two functional groups are present in the product



Two $3^{\circ}$ alcohol
Since the given question is single correct choice the best appropriate option is (A)
27. Official Ans. by NTA (1)

Sol.

28. Official Ans. by NTA (1)

Sol.

29. Official Ans. by NTA (3)

Sol.

(1) Aniline is lewis base give acid base reaction with $\mathrm{AlCl}_{3}$ and form Anilinium ion
(2) Anilinium ion has strongest deactivated ring so further Friedel craft Alkylation not occurs.
30. Official Ans. by NTA (3)

Sol. Bromine water gives tribromo products, other gives monobromo products in which para is major product.
31.

32. The density order

33. Official Ans. by NTA (2)

## Sol.



Yellow colored azo dye
34. Official Ans. by NTA (3)

Sol.


35. Official Ans. by NTA (2)

Sol.

36. Official Ans. by NTA (4)

Sol.


37. Official Ans. by NTA (2)

Sol. Only p-methyl, phenol does not give any colour with phthalic anhydroxide with cons. $\mathrm{H}_{2} \mathrm{SO}_{4}$.
38. Official Ans. by NTA (3)

Sol.


39. Official Ans. by NTA (1)

Sol.


40.

(i) $\mathrm{Sn}+\mathrm{HCl}$
(ii) $\mathrm{Fe}+\mathrm{HCl}$
(iii) $\mathrm{Zn}+\mathrm{HCl}$
(iv) $\mathrm{H}_{2}-\mathrm{Pd}$
(v) $\mathrm{H}_{2}$ (Raney Ni)
41. Official Ans. by NTA (1)

Sol.

(B)
42. Official Ans. by NTA (2)

43. Official Ans. by NTA (1)

Sol.

44. Official Ans. by NTA (1)

Sol.

[A]

