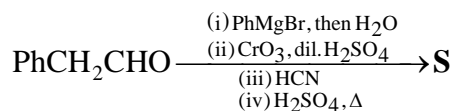
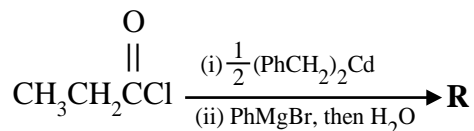
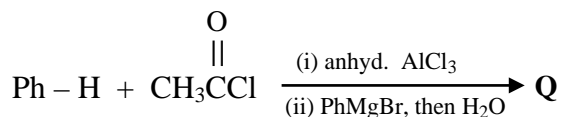
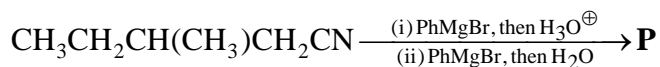


ORGANIC CHEMISTRY

CARBONYL COMPOUND

1. In the following reactions, **P**, **Q**, **R**, and **S** are the major products.

[JEE(Advanced) 2023]

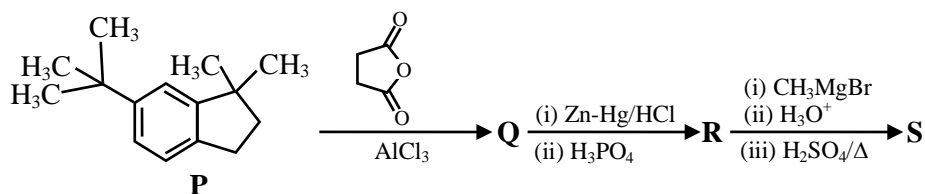


The correct statement(s) about **P**, **Q**, **R**, and **S** is(are)

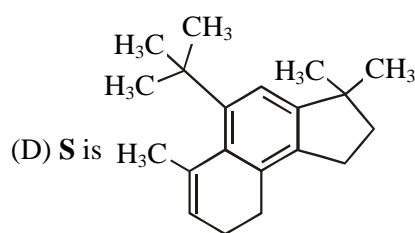
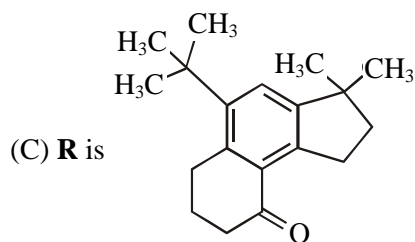
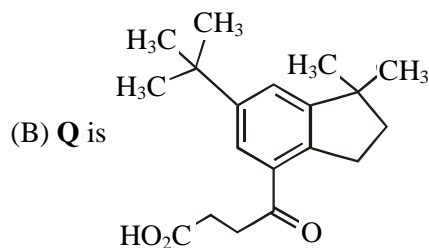
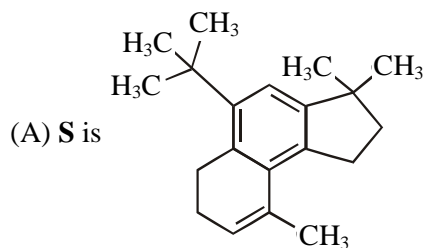
- (A) Both **P** and **Q** have asymmetric carbon(s).
- (B) Both **Q** and **R** have asymmetric carbon(s).
- (C) Both **P** and **R** have asymmetric carbon(s).
- (D) **P** has asymmetric carbon(s), **S** does **not** have any asymmetric carbon.

2. In the reaction scheme shown below **Q**, **R** and **S** are the major products.

[JEE(Advanced) 2020]

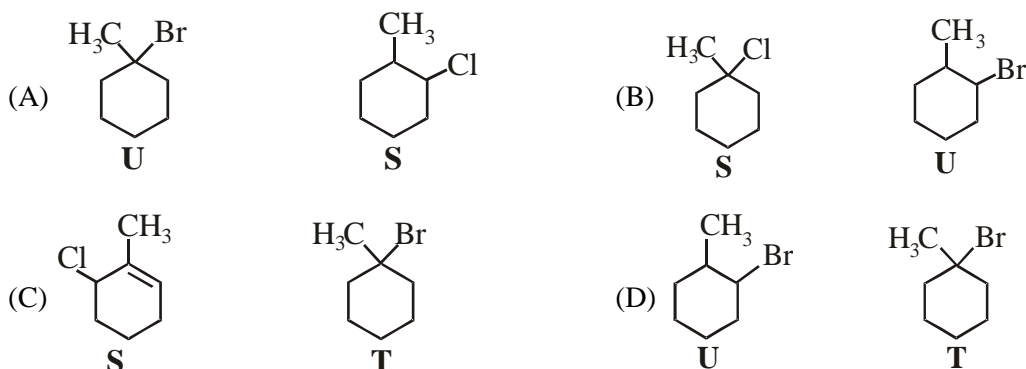
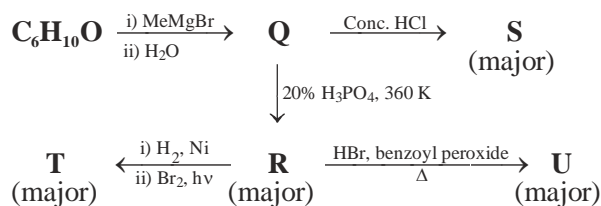


The correct structure of



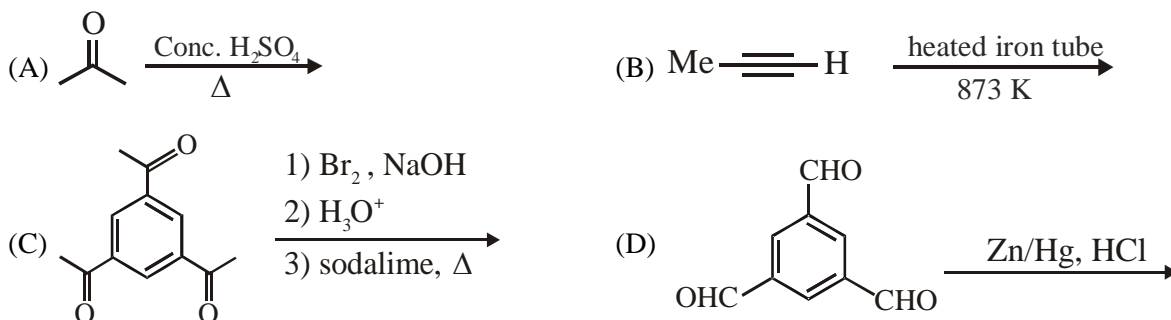
3. Choose the correct option(s) for the following set of reactions

[JEE(Advanced) 2019]



4. The reaction(s) leading to the formation of 1,3,5-trimethylbenzene is (are)

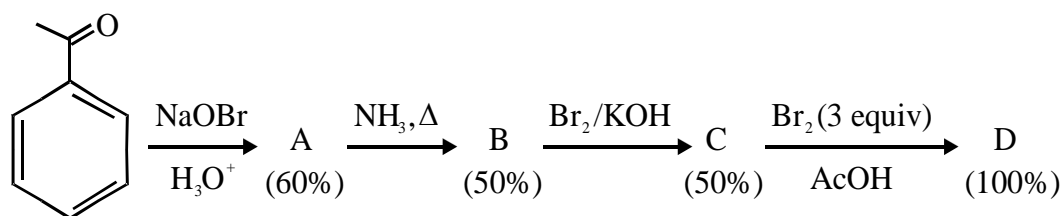
[JEE(Advanced) 2018]



5. In the following reaction sequence, the amount of D (in g) formed from 10 moles of acetophenone is \_\_\_\_\_.

(Atomic weight in  $\text{g mol}^{-1}$ : H = 1, C = 12, N = 14, O = 16, Br = 80. The yield (%) corresponding to the product in each step is given in the parenthesis)

[JEE(Advanced) 2018]



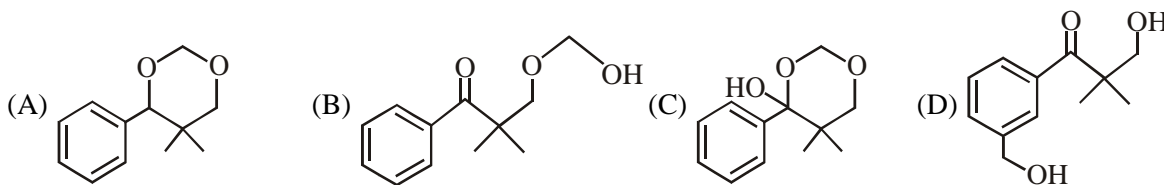
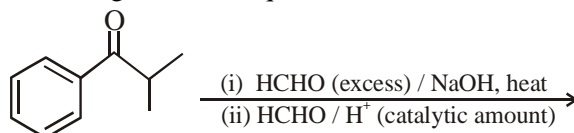
6. Positive Tollen's test is observed for

[JEE(Advanced) 2016]



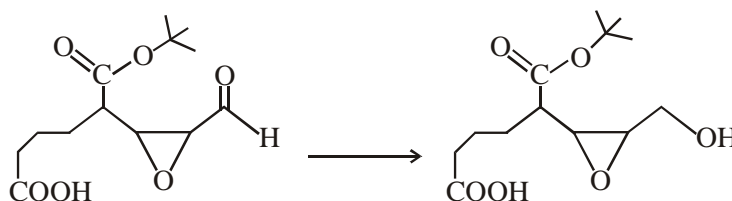
7. The major product of the following reaction sequence is :

[JEE(Advanced) 2016]



8. Reagent(s) which can be used to bring about the following transformation is(are)

[JEE(Advanced) 2016]



(A) LiAlH<sub>4</sub> in (C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>O

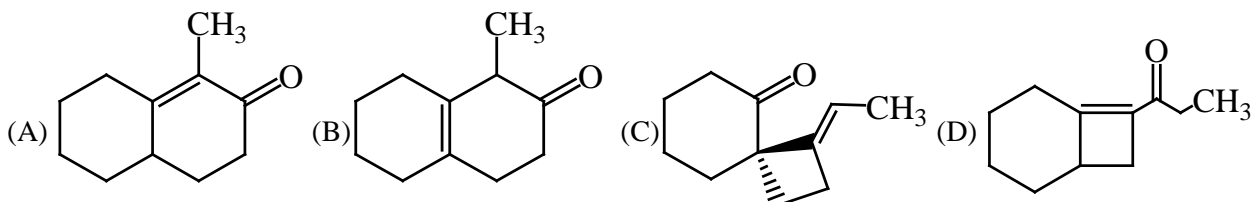
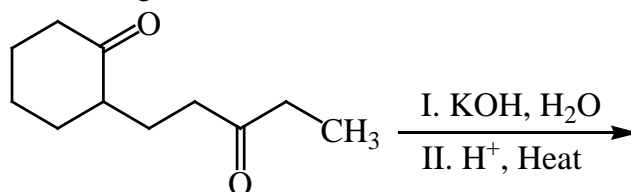
(B) BH<sub>3</sub> in THF

(C) NaBH<sub>4</sub> in C<sub>2</sub>H<sub>5</sub>OH

(D) Raney Ni / H<sub>2</sub> in THF

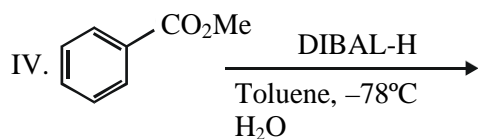
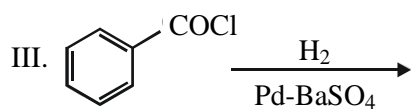
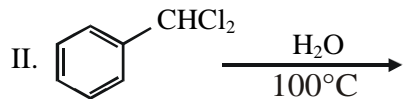
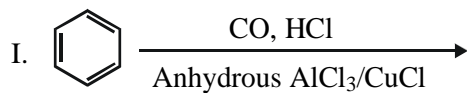
9. The major product of the following reaction is -

[JEE(Advanced) 2015]



10. Among the following the number of reaction(s) that produce(s) benzaldehyde is -

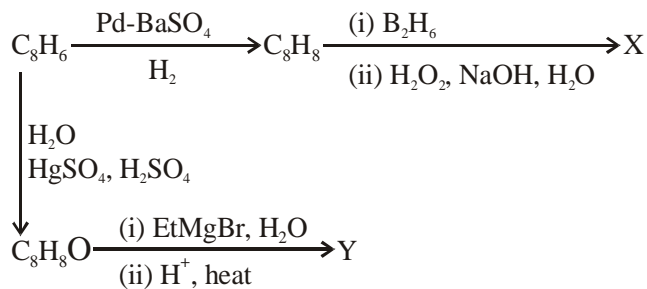
[JEE(Advanced) 2015]



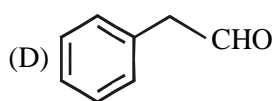
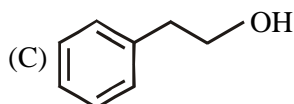
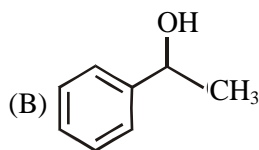
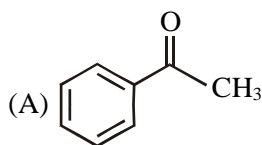
## Paragraph For Questions No. 11 and 12

In the following reaction

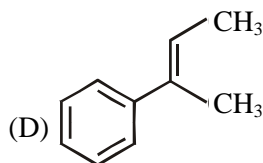
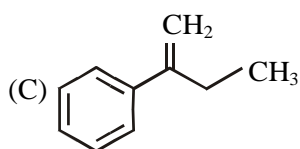
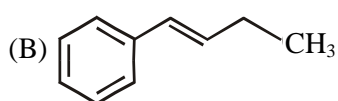
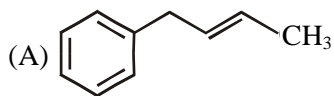
[JEE(Advanced) 2015]



11. Compound X is :



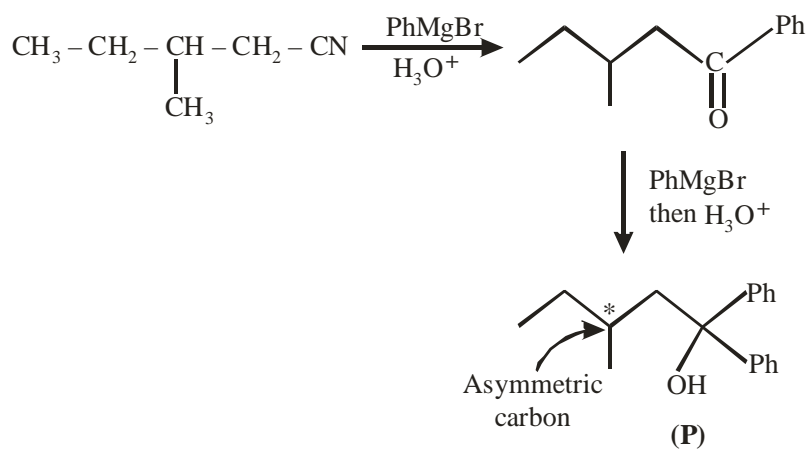
12. The major compound Y is :



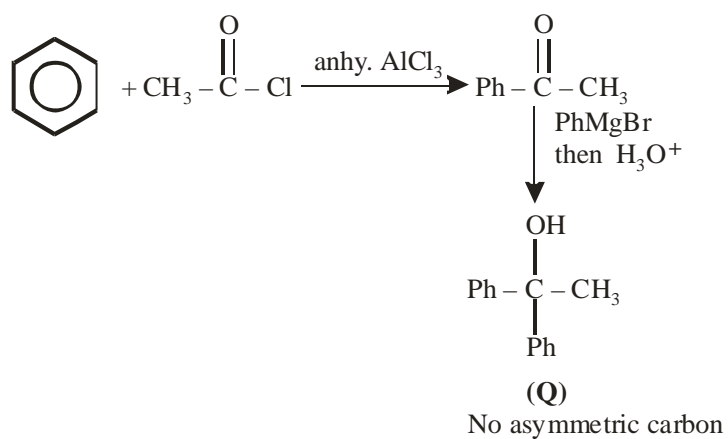
## SOLUTIONS

1. Ans. (C, D)

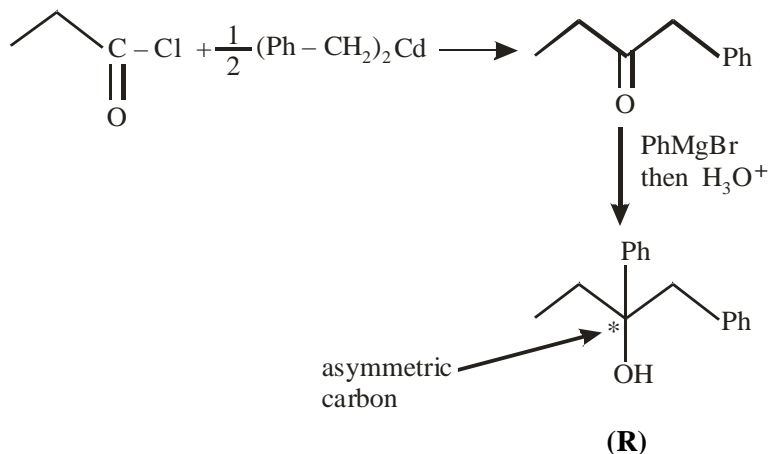
Sol. Formation of P



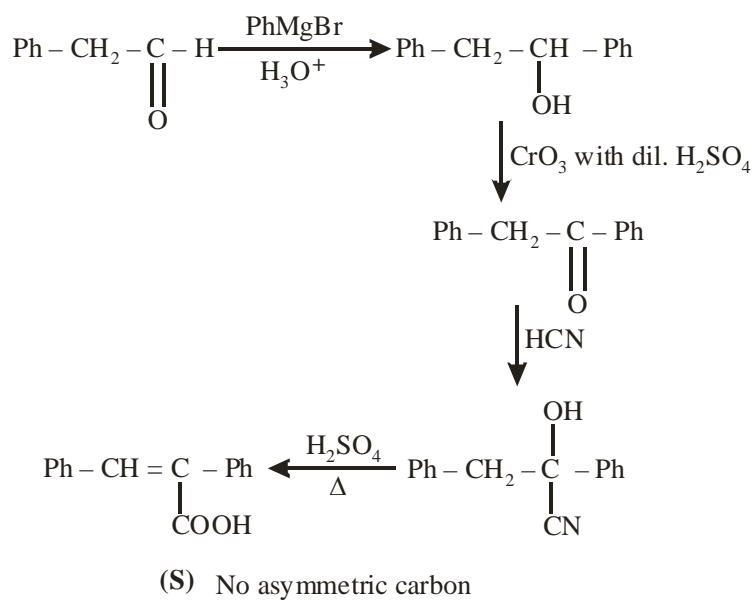
Formation of Q



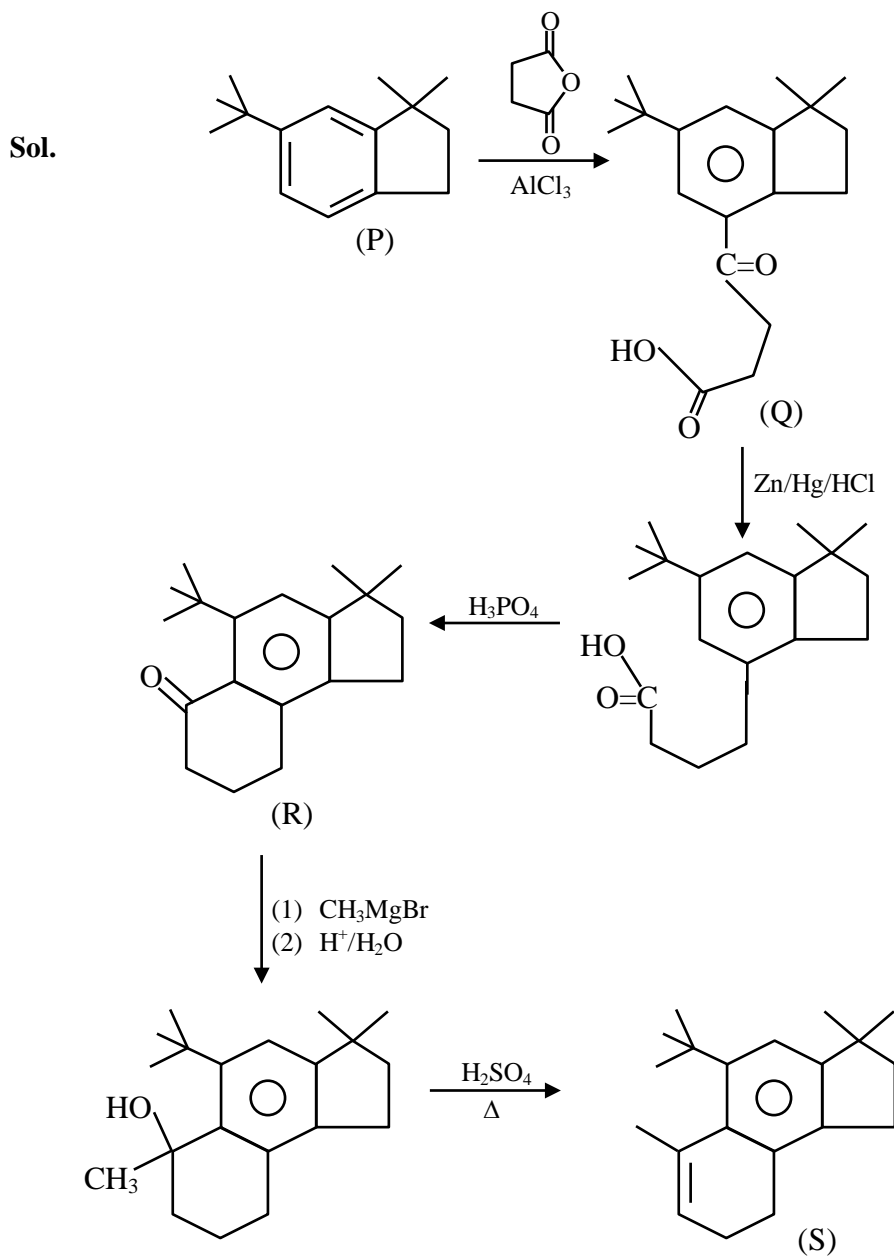
**Formation of R**



**Formation of S**

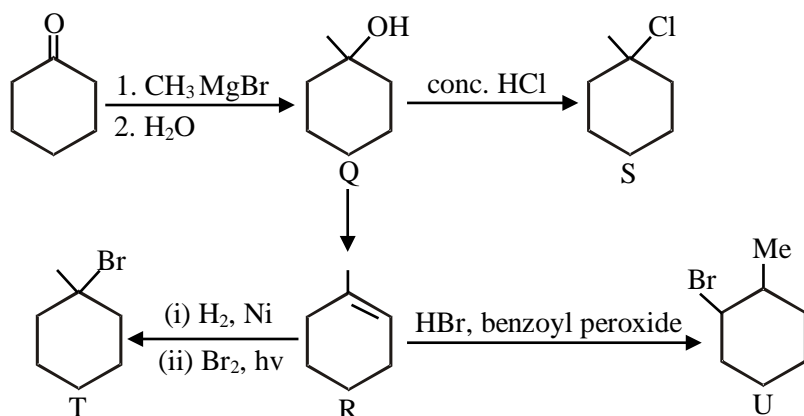


2. Ans. (B, D)

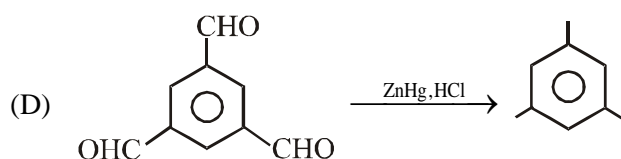
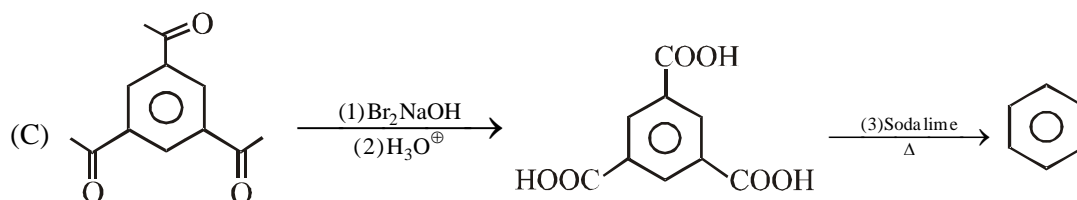
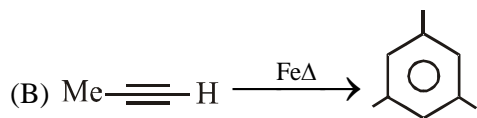
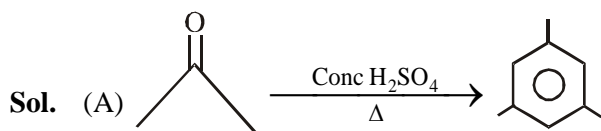


3. Ans. (B, D)

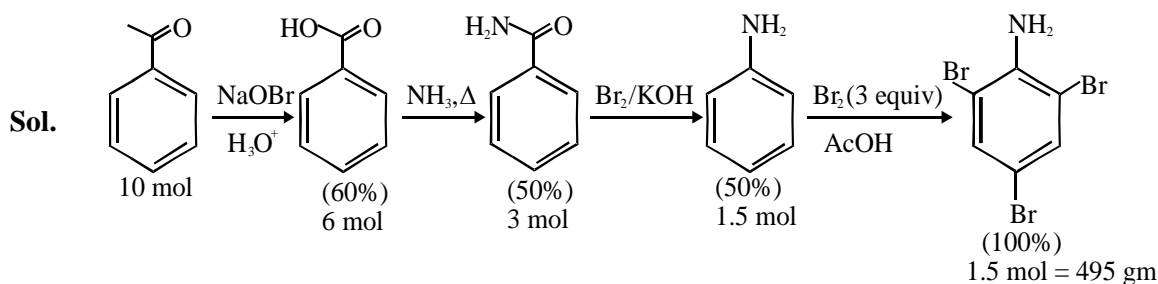
Sol.



4. Ans. (A, B, D)

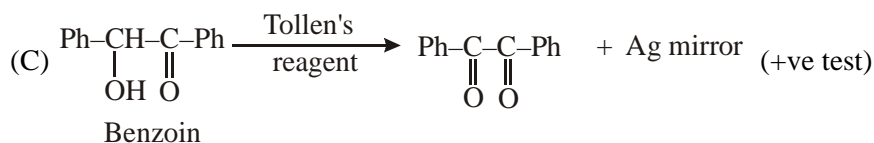
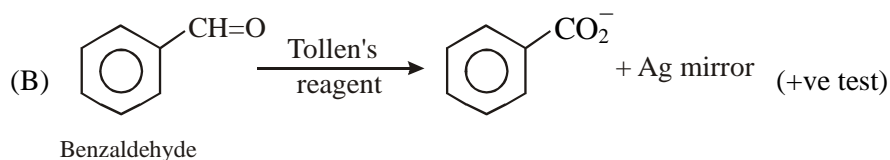
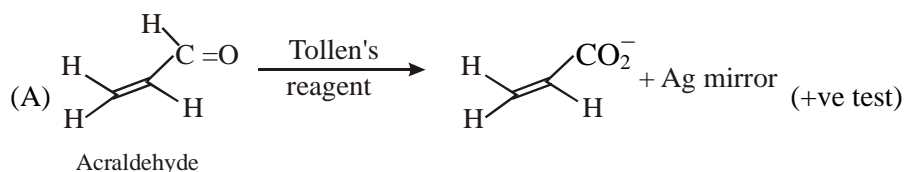


5. Ans. (495)



6. Ans. (A, B, C)

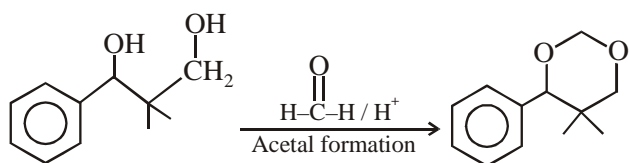
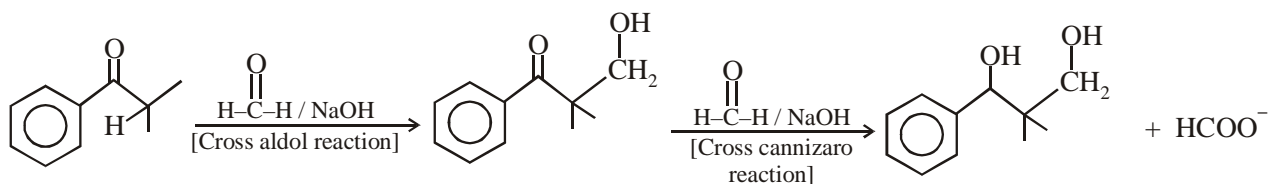
Sol. Tollens's test is given by compounds having aldehyde group. Also  $\alpha$ -hydroxy carbonyl gives positive tollen's test.





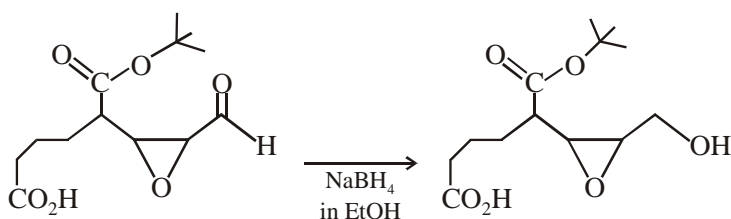
7. Ans. (A)

Sol.




8. Ans. (C, D)

Sol.



$\text{LiAlH}_4$  in  $(\text{C}_2\text{H}_5)_2\text{O}$ ;  $\text{BH}_3$  in (THF); Raney Ni ( $\text{H}_2$ ) either can reduce all functional group or can reduce some of the functional group of the compound given above in reactant side.

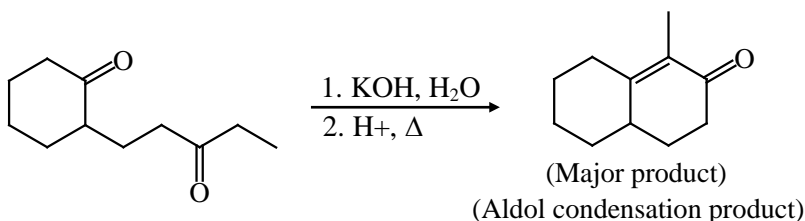
But  $\text{NaBH}_4$  is example of selective reducing agent. It can not reduce  $\text{-C(=O)-}$  (ester group)  $\text{-C(=O)OH}$ ,

(carboxylic acid group),  (epoxide group), but reduces  $\text{-CH=O}$  (aldehyde group) into  $\text{-CH}_2\text{OH}$

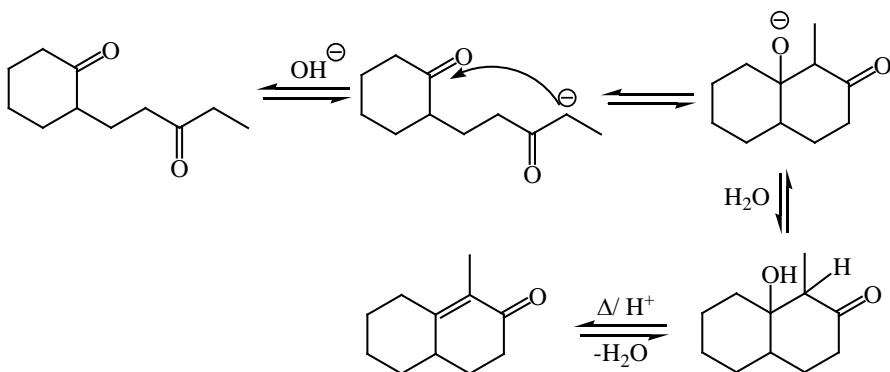
( $1^\circ$  alcohol)

9. Ans. (A)

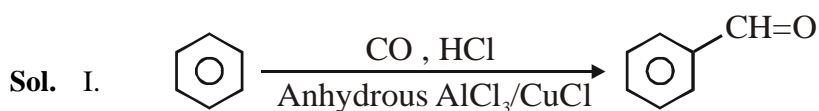
Sol.



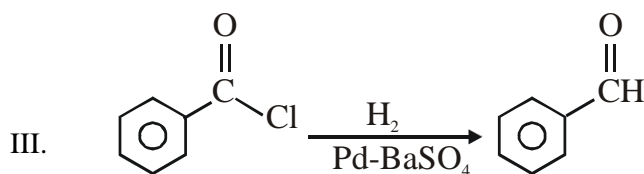
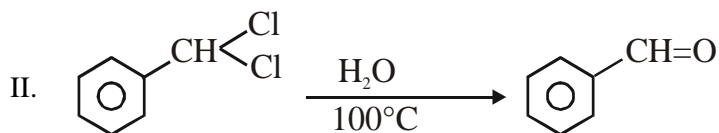
Mechanism :



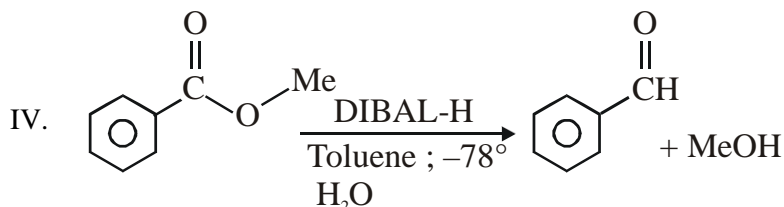
10. Ans. (4)



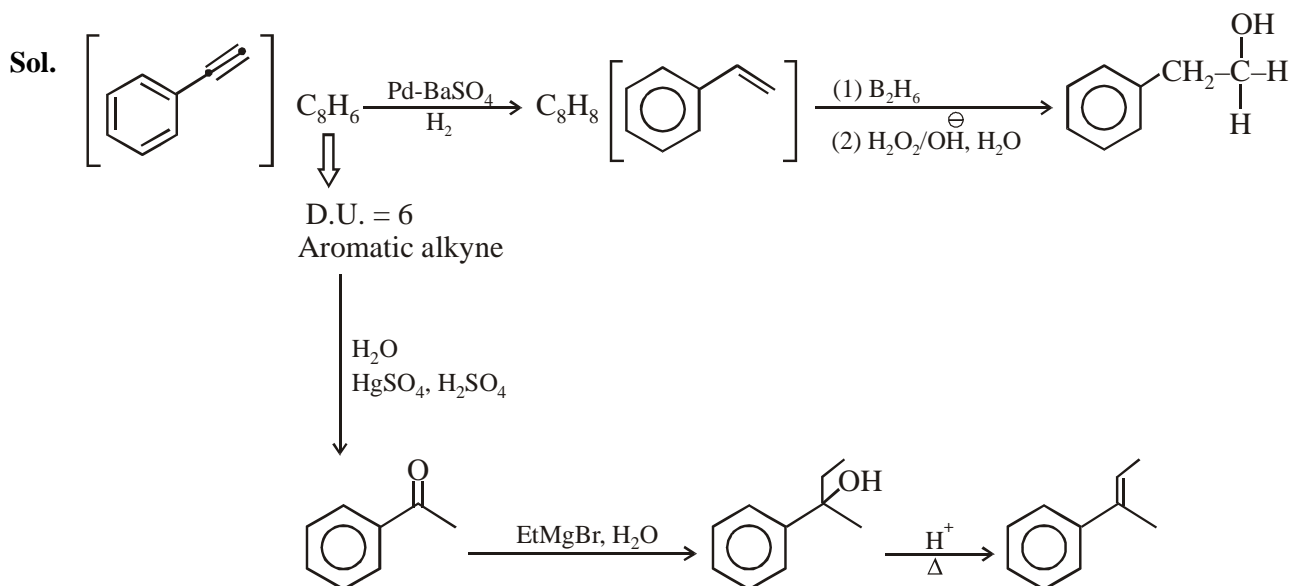
this reaction is called Gattermann koch synthesis



this reaction is called Rosenmund reduction



11. Ans. (C)



12. Ans. (D)

